

Pharmacokinetics

ACID-BASE, IONIZATION

WEAK ACIDS: aspirin, phenobarbital, penicillin, cephalosporins, loop & thiazide diuretics
WEAK BASES: morphine, local anaesthetics – ‘caines’, amphetamines, PCP

Weak Acid

$$\text{R-COOH} \rightleftharpoons \text{R-COO}^- + \text{H}^+$$

(crosses membranes) (better cleared)

Weak Base

$$\text{R-NH}_3^+ \rightleftharpoons \text{R-NH}_2 + \text{H}^+$$

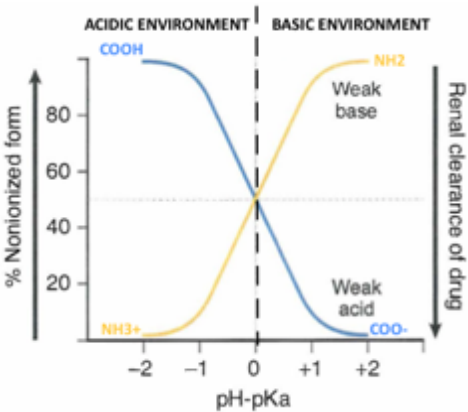
(better cleared) (crosses membranes)

Absorption/elimination of a weak acid drug

$$\text{pK}_a = \text{pH} - \log \frac{\text{concentration of ionized acid (A}^-\text{)}}{\text{concentration of nonionized acid (HA)}}$$

Absorption/elimination of a weak base drug

$$\text{pK}_a = \text{pH} - \log \frac{\text{concentration of nonionized base (B)}}{\text{concentration of ionized base (BH}^+\text{)}}$$



	PERCENTAGE IONIZATION	
pH - pKa	ACIDS	BASES
-3	0.1	99.9
-2	1	99
-1	10	90
0	50	50
1	90	10
2	99	1
3	99.9	0.1

A drug has a pKa of 8 and is administered to a patient with stomach pH of 2.
pH – pKa = 2-8 = -6 → This means that 99.9% of the drug is ionized & will not be absorbed in the stomach.

ELIMINATE the IONIZED form of the drug – water-soluble: Opposite environment of the drug

- Weak acid in BASIC ENVIRONMENT
- Weak base in ACIDIC ENVIRONMENT

ABSORB the NON-IONIZED form of the drug – lipid-soluble, can cross membranes: Same environment of the drug

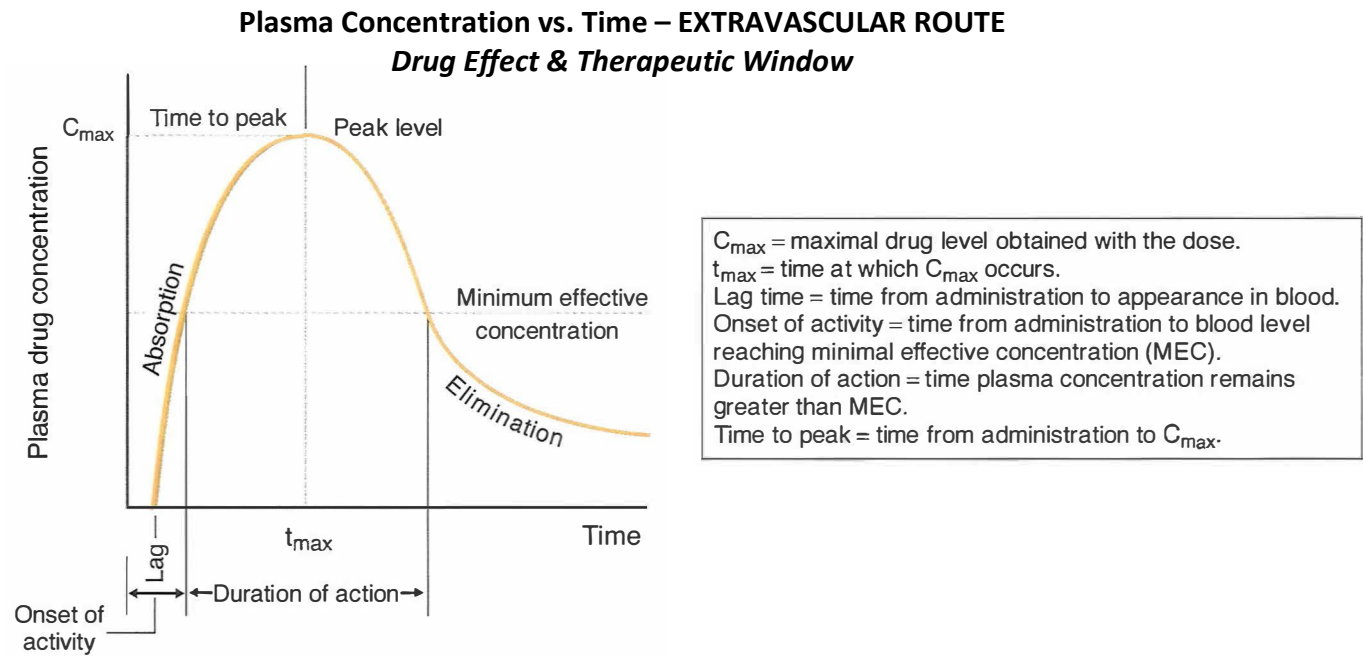
- Weak acid in ACIDIC ENVIRONMENT
- Weak base in BASIC ENVIRONMENT

To increase ionization (Elimination) of a Weak Base, ACIDIFY the urine – NH₄Cl, vitamin C, & cranberry juice
*i.e. OD on amphetamines

To increase ionization (Elimination) of a Weak Acid, ALKALINIZE the urine – NaHCO₃, acetazolamide
*i.e. OD on aspirin

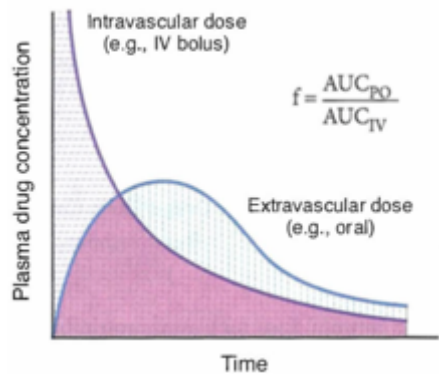
ABSORPTION: *Process of entry of a drug into the systemic circulation from the site of its administration*

**IV administration does NOT involve absorption & there is no loss of drug (bioavailability = 100%)
THUS, IV is NOT the fastest route of absorption.... INHALATION is the fastest route of absorption!*

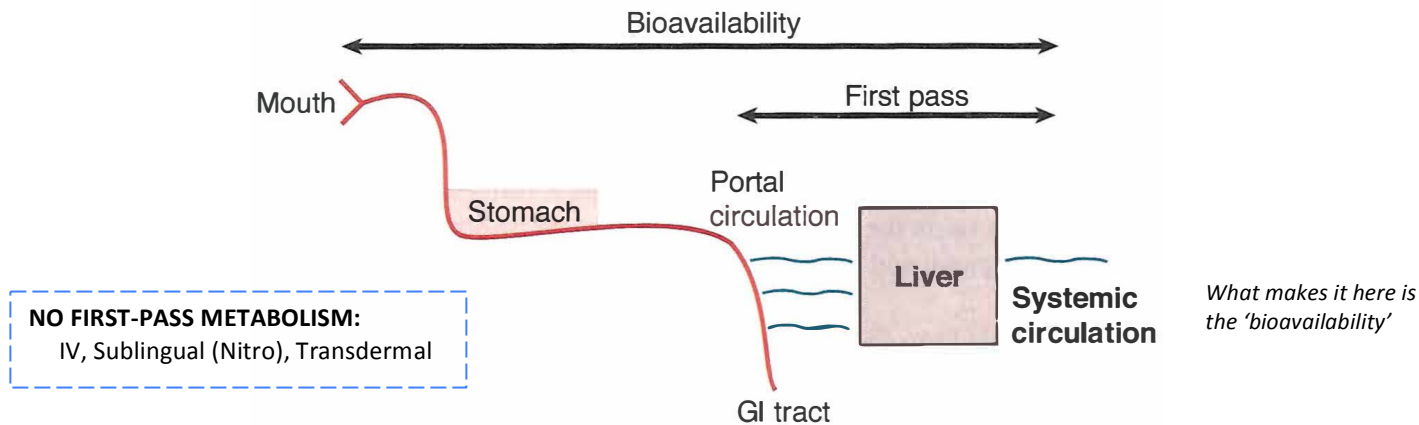


BIOAVAILABILITY: *Measure of the fraction of a dose that reaches the systemic circulation*

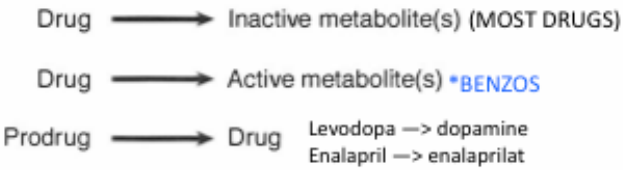
**BIOEQUIVALENT = same bioavailability + same plasma profile superimposed curves)*



FIRST-PASS EFFECT: *Drugs given PO are absorbed into Portal Circulation & initially distributed to LIVER*



BIOTRANSFORMATION (Metabolism): *Metabolic conversion of drugs to more water-soluble metabolites that are more readily excreted*



PHASE 1 REACTIONS: *Modification of the drug molecule via oxidation, reduction, or hydrolysis*

- Cytochrome P450 isoenzymes are major enzymes involved in phase I reactions
- **INDUCERS:** ↓ substrate level of drug
- **INHIBITORS:** ↑ substrate level of drug
 - o **GRAPEFRUIT JUICE:** inhibits 3A4, which inhibits statin metabolism & thus, increases statin concentration in plasma

Table I-1-2. Cytochrome P450 Isozymes

CYP450	Substrate Example	Inducers	Inhibitors	Genetic Polymorphisms
1A2	Theophylline Acetaminophen	Aromatic hydrocarbons (smoke) Cruciferous vegetables	Quinolones Macrolides	No
2C9	Phenytoin Warfarin	General inducers*	—	Yes
2D6	Many cardiovascular and CNS drugs	None known	Haloperidol Quinidine	Yes
3A4	60% of drugs in PDR	General inducers*	General inhibitors† Grapefruit juice	No

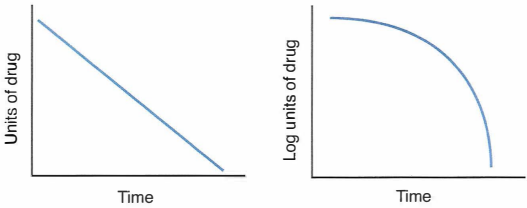
* General inducers: anticonvulsants (barbiturates, phenytoin, carbamazepine), antibiotics (rifampin), chronic alcohol, St. John's Wort.
† General inhibitors: antiulcer medications (cimetidine, omeprazole), antimicrobials (chloramphenicol, macrolides, ritonavir, ketoconazole), acute alcohol.

PHASE II REACTIONS: *Conjugation with endogenous compounds via the activity of **transferases***

1. **Glucuronidation:** Diazepam, Digoxin, Morphine, Chloramphenicol
 - CHLORAMPHENICOL TOXICITY: Grey Baby Syndrome due to low levels of glucuronosyl transferase
2. **Acetylation:** Sulfonamides, INH
 - Genotypic variations
 - Drug-Induced SLE by slow acetylators with hydralazine: *Stop drug & SLE goes away*
3. **Glutathione:** GSH
 - Depletion of GSH in the liver associated with acetaminophen hepatotoxicity

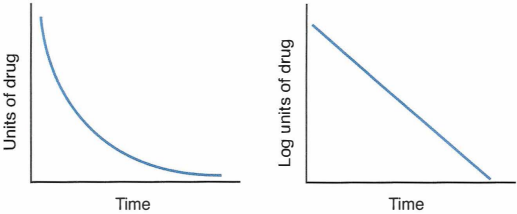
ZERO ORDER KINETICS: *Think ‘saturation kinetics’*

- Fixed amount of drug eliminated per unit time
 - o **RATE OF ELIMINATION = CONSTANT**
- **Independent** of plasma concentration
- Variable half-life
- **Phenytoin, Ethanol, Aspirin, Salicyclates – PEAS***



FIRST ORDER KINETICS

- Constant fraction of drug eliminated per unit time
- Concentration dependent: *rate falls as plasma levels fall*
- **Clearance is CONSTANT & HALF-LIFE is CONSTANT**
- *Most drugs*



Rate of administration
$$R_A = \frac{\text{dose} \times (\text{bioavailability } F \times \text{salt form } S)}{\text{dosing interval } \tau}$$

*V_d is LOW when high % of drug is bound to plasma protein (↑C_p)
V_d is HIGH when high % of the drug is sequestered in tissues (↓C_p)*

Volume of distribution
$$V_d = \frac{\text{total amount of drug in body}}{\text{plasma concentration}}$$

$$= \frac{\text{dose} \times (\text{bioavailability } F \times \text{salt form } S)}{\text{plasma concentration } C_p}$$

**If V_d is given in L/kg, you have to multiply by the patient's weight – if no weight given, assume 70kg*

****Rule of Thumb:** Loading dose = 2 x Maintenance Dose
This is how it will be on the exam!*

Loading dose
$$\text{Loading Dose} = (C_p \times V_d) / (F \times S)$$

Loading dose (maintenance form)
$$LD = 1.44 \times t_{1/2} / \tau \times MD$$

# of half-lives	1	2	3	4
% remaining	50%	25%	12.5%	6.25%

Half-life *ONLY FOR FIRST ORDER KINETICS!*
$$t_{1/2} = \frac{0.693 \times V_d}{Cl}$$
 or
$$t_{1/2} = \frac{0.693}{k_e}$$

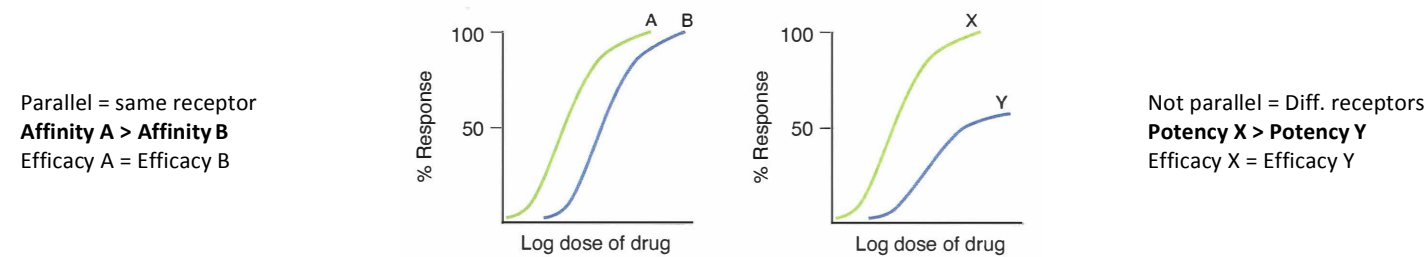
% drug remaining after t time
$$(0.5)^{t/t_{1/2}}$$

Clearance
$$Cl = \frac{0.693 \times V_d}{t_{1/2}}$$
 or
$$Cl = k_e \times V_d$$

Concentration of a drug at steady state
$$C_{ss} = \frac{\text{rate of administration } R_A}{\text{clearance } Cl}$$

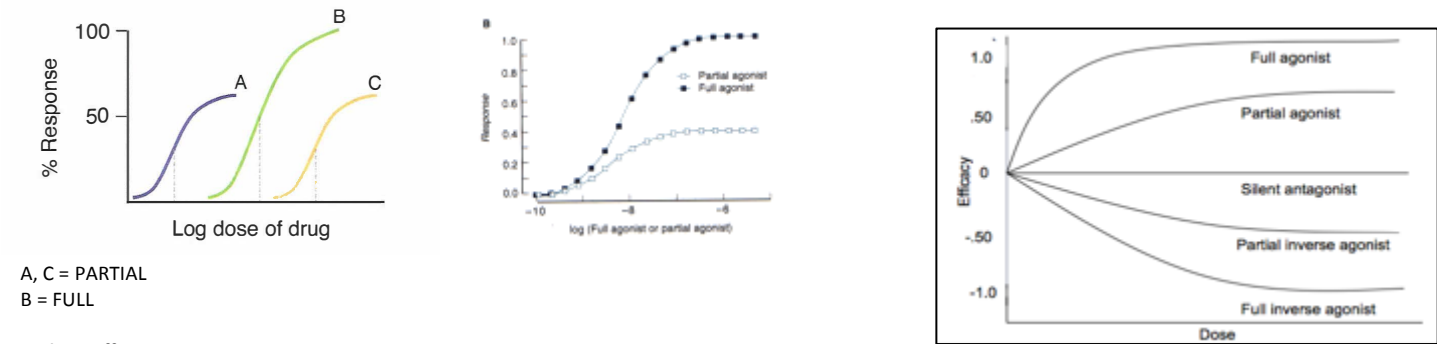
Pharmacodynamics

AFFINITY: the nearer the curve to the y-axis, the greater the affinity; parallel curves
POTENCY: shows relative doses of 2+ agonists to produce the *same magnitude of effect* can compare if 2 curves do not cross
EFFICACY: effectiveness; the dose required to produce a certain effect; shown by max height reached by the curve;



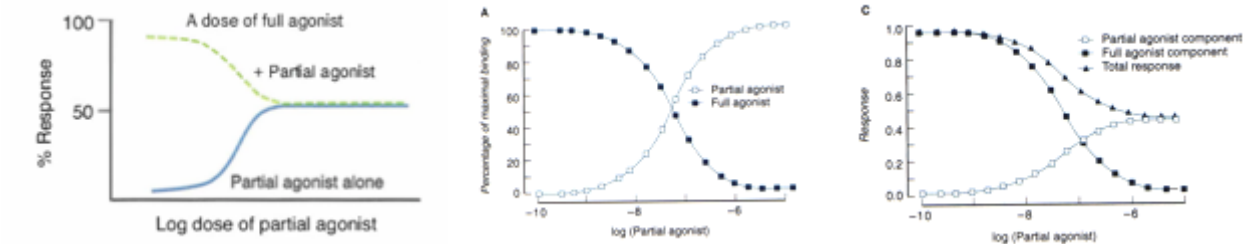
AGONIST: when binding to the receptor gives a response (*full, partial inverse*)

- FULL AGONIST: produce a maximal response; maximal efficacy
- PARTIAL AGONIST: incapable of eliciting maximal response; less effective
 - o Can act as an antagonist – competition for receptor
- INVERSE AGONIST: produces an opposite effect of the agonists; *stabilizes the inactive state reducing signal transduction below basal levels*



A, C = PARTIAL
B = FULL

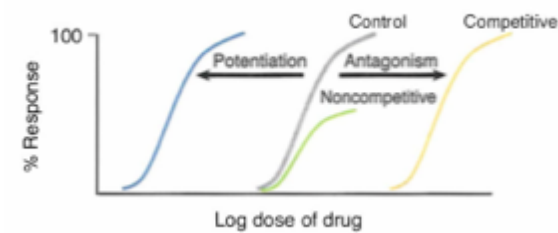
Highest efficacy: B
Highest potency: A (A>C; B>C)
Can't compare potency of A & B because their curves cross, but you can compare at a specific dose. At low doses, A>B & at high doses, B>A.



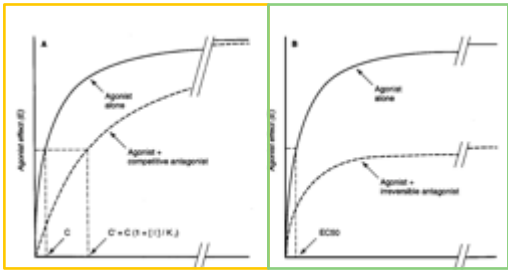
As the partial agonist replaces the full agonist from the receptors, the receptor is reduced – *Partial agonist is acting as ANTAGONIST*

Example: Epinephrine, Pindolol (β -blocker, partial agonist), & heart beat

ANTAGONIST: binding to the receptor is NOT associated with a response; the drug only has an effect by **preventing** an agonist from binding the receptor; *stabilizes the equilibrium condition*

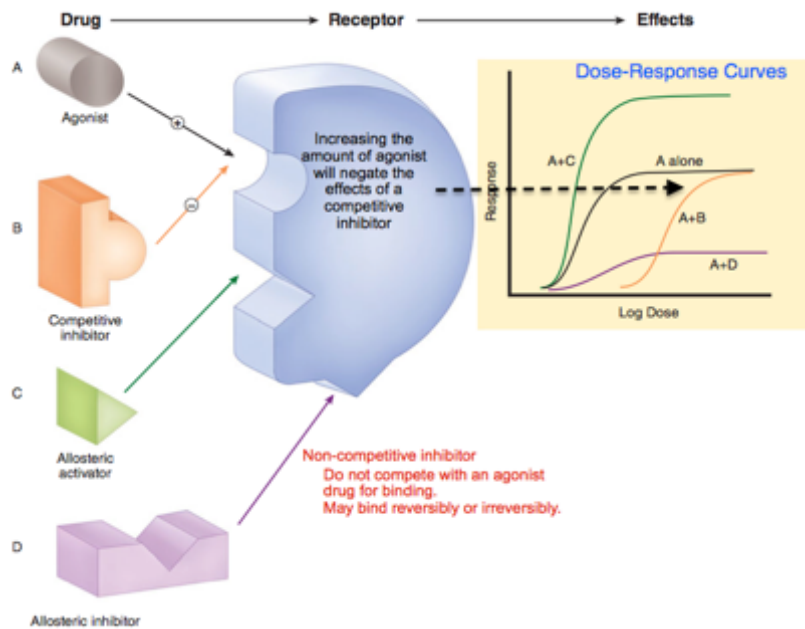


COMPETITIVE: ↓ **POTENCY** (shift right, ΔED_{50})
POTENTIATION: ↑ **POTENCY** (shift left)
NON-COMPETITIVE: ↓ **EFFICACY** (shift down)

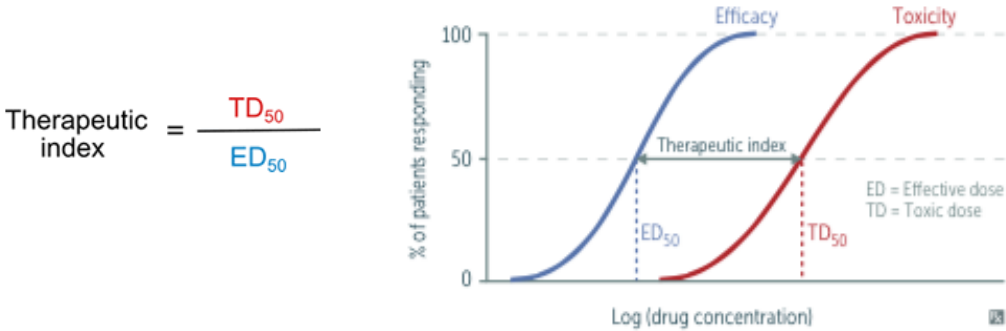


TYPES OF ANTAGONISM:

- **PHARMACOLOGIC: SAME RECEPTORS**
 - **COMPETITIVE:** *Atropine, propranolol*
 - Occupies receptor of agonist without activating it
 - Parallel shift in dose-response curve for agonist
 - Overcome with increasing dose of agonist
 - **NON-COMPETITIVE:** *Aspirin, phenoxybenzamine*
 - Non-parallel shift to right
 - Can only be partially overcome with increasing dose of agonist
- **PHYSIOLOGIC: DIFFERENT RECEPTORS**
 - Two agonists with **OPPOSING** actions antagonize each other
 - *Phenylephrine (vasoconstrictor) + Nitroglycerin (vasodilator)*
- **CHEMICAL: NO RECEPTORS (Acts directly with the agonist)**
 - Formation of a complex between effector drug & another compound
 - *Protamine binds to heparin to reverse its actions*



- THERAPEUTIC INDEX:** how effective a drug is in producing the desired effect relative to toxicity
- Both ED50 and TD50 are calculated from quantal-dose response curves, which represent the frequency with which each dose of drug elicits the desired response or toxic effect in the population
 - **ED50:** dose required to produce a therapeutic effect in 50% of the population
 - **TD50:** dose required to produce a toxic effect in 50% of the population
 - **SAFER DRUGS HAVE HIGHER TI VALUES**
 - **DRUGS WITH NARROW/LOW THERAPEUTIC INDEX:** Warfarin, Lithium, Digoxin, Theophylline, Gentamycin, Amphotericin B



Drug Development

Pre-clinical	Phase 1	Phase 2	Phase 3	Phase 4
Animals Safety & biologic activity	~20 Healthy volunteers <i>*THIS IS NOT PATIENTS</i> Safety, minimum effective dose (pharmacokinetics)	100s Selected PATIENTS <i>Therapeutic efficacy</i>	Large # Selected PATIENTS Safety & efficacy Overall benefit-risk ratio <i>*SIDE EFFECTS</i>	Post-marketing surveillance – Patients given drug for therapy Adverse reactions

IND application for
FDA Safety Review

NDA Review